

**Answer ALL SIX questions from Section A and THREE questions from Section B.**

The numbers in square brackets at the right-hand margin indicate the provisional allocation of marks per sub-section of a question.

You may assume the following values:

### SECTION A

[Part  
marks]

1. What relationship did de Broglie postulate between the wavelength of a matter-wave and the momentum of the corresponding particle? [2]

The function

$$\psi(x) = A \exp(ikx)$$

is a solution to the time-independent Schrödinger equation for a free particle of mass  $m$  moving in one dimension. How is  $k$  related to the de Broglie wavelength? [1]

If the potential energy  $V$  is zero everywhere, substitute the state  $\psi$  into the Schrödinger equation to find the total energy  $E$ . Show that your result gives the kinetic energy, assuming de Broglie's hypothesis is correct. [3]

2. Write down the time-*dependent* Schrödinger equation for a particle of mass  $m$  moving in one dimension in a potential  $V$  and wavefunction  $\Psi(x, t)$ . [2]

Assuming that the potential  $V$  is independent of time, find the equation satisfied by the function  $\psi(x)$  such that

$$\Psi(x, t) = \psi(x) \exp(-iEt/\hbar)$$

is a solution of the time-dependent Schrödinger equation. [3]

Show that, if  $\Psi_1$  and  $\Psi_2$  are both solutions of the time-dependent Schrödinger equation, so is  $c_1\Psi_1 + c_2\Psi_2$  (where  $c_1$  and  $c_2$  are arbitrary constants). [2]

3. The expansion postulate says that any function  $\psi(x)$  can be expanded in terms of the complete set of eigenfunctions  $\phi_n(x)$  of a Hermitian operator  $\hat{O}$  in the form

$$\psi(x) = \sum_n c_n \phi_n(x).$$

If the  $\phi_n$  form an orthonormal set and  $\psi$  is normalized so

$$\int_{-\infty}^{\infty} |\psi(x)|^2 dx = 1,$$

what (according to the postulates of quantum mechanics) is the connection between the coefficients  $c_n$  and the probabilities of obtaining different results when the physical quantity corresponding to  $\hat{O}$  is measured? (A derivation is *not* required.) [3]

Show that in these circumstances the expansion coefficients  $c_n$  must satisfy the condition

$$\sum_n |c_n|^2 = 1. \quad [3]$$

4. The spherical harmonics  $Y_l^m(\theta, \phi)$  are simultaneously eigenfunctions of which two angular momentum operators? [2]

Give the corresponding eigenvalues of the two operators in terms of the quantum numbers  $l$  and  $m$ . [2]

For a given  $l$ , what are the allowed values of  $m$ ? Give an interpretation of the range of allowed  $m$  values within the vector model of angular momentum. [4]

5. Far from the nucleus, and in atomic units, the radial part of the Schrödinger equation for a hydrogen atom can be written

$$-\frac{1}{2} \frac{d^2\chi}{dr^2} = E\chi.$$

For a bound state ( $E < 0$ ), write down two independent solutions of this equation in terms of the quantity  $\kappa = \sqrt{2|E|}$ . [2]

Which one of your two solutions is physically admissible, and why? [2]

What is the energy of the ground state of a hydrogen atom in atomic units? Give also the corresponding value of  $\kappa$ , in atomic units. [2]

6. When spin is introduced into quantum physics, the state of an electron is characterized by two new quantum numbers:  $s$  and  $m_s$ . What is the physical significance of these quantum numbers? [2]

One of these quantum numbers always takes the same value for all states of a single electron; which is it? [1]

Briefly describe *one* piece experimental evidence for the accepted value of this quantum number. [4]

## SECTION B

7. A particle of mass  $m$  moves in one dimension in an infinite square well extending from  $x = -a$  to  $x = +a$ . Inside the well the potential is zero. Write down the time-independent Schrödinger equation for the particle when it is inside the well, and find the most general solution in this region. [5]

What are the boundary conditions on the wave function at  $x = \pm a$ ? By applying these boundary conditions, find the energies and wavefunctions for the square well's *even* stationary states, i.e. those having  $\psi(x) = \psi(-x)$ . [5]

For the case of the ground state (the lowest-energy solution) normalize the wavefunction so that

$$\int_{-\infty}^{\infty} |\psi|^2 dx = 1.$$

[3]

The *expectation value* of an operator  $\hat{O}$  in a quantum system with wavefunction  $\psi(x)$  can be written as

$$\langle \hat{O} \rangle = \int_{-\infty}^{\infty} \psi^*(x) \hat{O} \psi(x) dx.$$

What is the significance of the expectation value in relation to the results of measurements of  $\hat{O}$ ? [2]

Using your normalized wavefunction, calculate the expectation value in the ground state of (i) momentum  $\hat{p}$ , (ii) position  $\hat{x}$ , (iii) position squared  $\hat{x}^2$ . [5]

*You may use without proof the results*

$$\int_{-\pi/2}^{\pi/2} \cos^2 \theta d\theta = \frac{\pi}{2}; \quad \int_{-\pi/2}^{\pi/2} \theta^2 \cos^2 \theta d\theta = \frac{\pi}{24}(\pi^2 - 6).$$

8. A beam of particles with mass  $m$  and energy  $E$  is incident from the left on a potential step of height  $V_0$  such that

$$V(x) = \begin{cases} 0 & (x \leq 0) \\ V_0 & (x > 0). \end{cases}$$

Sketch the potential  $V(x)$ . [1]

Assuming that  $E > V_0 > 0$  and that the incident beam is normalized so it contains one particle per unit length, write down appropriate solutions for the time-independent Schrödinger equation (a) for  $x \leq 0$ , (b) for  $x > 0$ , in terms of the quantities  $k$  and  $k'$  defined by

$$\frac{\hbar^2 k^2}{2m} = E; \quad \frac{\hbar^2 k'^2}{2m} = E - V_0.$$

Pay special attention to which terms in your solutions are allowed by the boundary conditions. [4]

Write down the matching conditions that must be satisfied by the wavefunction at  $x = 0$ , and use them to find the unknown coefficients in your wave-function. [6]

By using your results to compare the probability fluxes in the reflected and incident beams, show that the reflection probability for the particles from the potential step is

$$R = \frac{(k - k')^2}{(k + k')^2}. \quad \text{[3]}$$

For the case where the incident energy is  $E = 4V_0/3$ , find  $k'$  in terms of  $k$ . Hence show that to the left of the step ( $x \leq 0$ ) the probability per unit length for finding a particle is  $10/9 + 2 \cos(2kx)/3$ . Also find an expression for the probability per unit length on the right of the step ( $x > 0$ ). [6]

[For a stationary state  $\psi(x)$  in one dimension, the probability flux is

$$\Gamma(x) = \frac{-i\hbar}{2m} \left[ \psi^* \frac{\partial \psi}{\partial x} - \psi \frac{\partial \psi^*}{\partial x} \right] = \frac{\hbar}{m} \text{Im} \left[ \psi^* \frac{\partial \psi}{\partial x} \right]$$

at position  $x$ .]

9. Starting from the classical definition of angular momentum  $\mathbf{L} = \mathbf{r} \times \mathbf{p}$ , show that the quantum mechanical operators for the  $x$ - and  $z$ -components of angular momentum are

$$\hat{L}_x = -i\hbar \left[ y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right]; \quad \hat{L}_z = -i\hbar \left[ x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right]. \quad [4]$$

Verify that the state  $\psi = Cxf(r)$ , where  $C$  is a constant and  $f$  is any function of  $r = \sqrt{x^2 + y^2 + z^2}$ , is an eigenfunction of  $\hat{L}_x$ , and find the corresponding eigenvalue. [3]

Find the angular part  $g(\theta, \phi)$  of the function  $\psi$ , using the normal definition of spherical polar coordinates, and show that if we choose  $C = \sqrt{\frac{3}{4\pi}}$  then  $g$  is normalized so

$$\int_0^{2\pi} d\phi \int_0^\pi \sin \theta d\theta |g|^2 = 1. \quad [5]$$

The completeness postulate implies that  $g$  can be expanded in terms of the spherical harmonics

$$g(\theta, \phi) = \sum_{lm} a_{lm} Y_l^m(\theta, \phi),$$

where the  $a_{lm}$  are constant expansion coefficients. By inspecting your expression for  $g(\theta, \phi)$  and comparing with the expressions given below for the spherical harmonics  $Y_l^m$ , find these expansion coefficients. [4]

Hence calculate the possible results of measuring the operators  $\hat{L}^2$  and  $\hat{L}_z$  in the state  $\psi$ , and the corresponding probabilities. [4]

[You may assume the results

$$\frac{\partial r}{\partial y} = \frac{y}{r}; \quad \frac{\partial r}{\partial z} = \frac{z}{r}; \quad \int_0^\pi \sin^3 \theta d\theta = \frac{4}{3}; \quad \int_0^{2\pi} \cos^2 \phi d\phi = \pi,$$

and the expressions for the following spherical harmonics  $Y_l^m$ :

$$Y_1^0 = \sqrt{\frac{3}{4\pi}} \cos \theta; \quad Y_1^1 = -\sqrt{\frac{3}{8\pi}} \sin \theta \exp(i\phi); \quad Y_1^{-1} = \sqrt{\frac{3}{8\pi}} \sin \theta \exp(-i\phi).$$

Remember that  $\cos \phi = [\exp(i\phi) + \exp(-i\phi)]/2.$

10. The time-independent Schrödinger equation for the electron in a hydrogen atom, in atomic units, is

$$-\frac{1}{2}\nabla^2\psi - \frac{1}{r}\psi = E\psi.$$

If a solution  $\psi$  is written in spherical polar coordinates  $(r, \theta, \phi)$  as

$$\psi(r, \theta, \phi) = R(r)Y(\theta, \phi),$$

show by substituting into the Schrödinger equation that the angular function  $Y$  is an eigenfunction of the total (orbital) angular momentum  $\hat{L}^2$ . [5]

If we neglect spin and all relativistic effects, the allowed energies are determined by a single principal quantum number  $n$ . What are the allowed values of  $n$ , and what range of  $l$  values is found for each value of  $n$ ? [3]

Hence show, again neglecting spin and using the formula for the sum of an arithmetic progression given below, that the total number of solutions to the time-independent Schrödinger equation with the same energy (and hence the same value of  $n$ ) is

$$\sum_{l=0}^{l=n-1} (2l + 1) = n^2. \quad [3]$$

How does this total number of states change when the electron's spin is accounted for? [2]

Including spin, the change in the atom's Hamiltonian when an external magnetic field  $(0, 0, B)$  applied in the  $z$ -direction is

$$\Delta\hat{H} = \frac{\mu_B B}{\hbar}(\hat{L}_z + g\hat{S}_z),$$

where  $\mu_B$  is the Bohr magneton,  $\hat{L}_z$  and  $\hat{S}_z$  are respectively the  $z$ -components of the orbital and spin angular momentum of the electron. Taking  $g = 2$ , explain how the 2p energy levels of a hydrogen atom are split in the presence of such a magnetic field. [5]

What are the possible values of the total angular momentum quantum number  $j$  that could be obtained by combining the spin and orbital angular momenta for such a 2p electron? [2]

[The Laplacian operator may be written in spherical polar coordinates (using atomic units) as

$$\nabla^2 = \frac{1}{r^2} \frac{d}{dr} \left[ r^2 \frac{d}{dr} \right] - \frac{\hat{L}^2}{r^2}.$$

The sum of  $N$  terms of an arithmetic progression having first term  $a$  and last term  $b$  is  $N(a + b)/2$ .]

11. State the relationship between the eigenfunctions and eigenvalues of an operator  $\hat{O}$ , and explain their role in quantum mechanics when the physical quantity corresponding to  $\hat{O}$  is measured. [5]

Two operators  $\hat{Q}$  and  $\hat{R}$  are said to be *compatible* if they share the same eigenfunctions  $\{\phi_n\}$ . Use the expansion postulate applied to an arbitrary function  $\psi$  to show that

$$[\hat{Q}, \hat{R}]\psi = 0$$

for any pair of compatible operators. [4]

According to the measurement postulate of quantum mechanics, what happens if  $\hat{Q}$  is measured, followed immediately by a measurement of a compatible quantity  $\hat{R}$  and then a further measurement of  $\hat{Q}$ ? (You may assume that the eigenvalues of  $\hat{Q}$  and the eigenvalues of  $\hat{R}$  are all different.) [4]

Which pairs of operators from the set  $\{\hat{L}^2, \hat{L}_x, \hat{L}_z\}$  are compatible with one another? [2]

An electron in a hydrogen atom is initially in the state

$$\psi = \frac{1}{\sqrt{2}}(\psi_{2,1,1} + \psi_{2,0,0}),$$

where  $\psi_{n,l,m}$  denotes the normalized wavefunction with quantum numbers  $n$ ,  $l$  and  $m$ . What happens if first  $\hat{L}^2$  is measured, then  $\hat{L}_z$ , and finally  $\hat{L}^2$  again? At each stage, give the possible outcomes of the measurement and the associated probabilities, and the state in which the system is left after the measurement. [5]